

## Thermal Conductivity: Drude model

Drude model's most impressive success (at the time) was the explanation of the Wiedemann-Franz law

### Wiedemann-Franz law:

$$\frac{\text{thermal conduct.}}{\text{electrical conduct.}} = \frac{K}{\sigma} \propto T \quad \left. \vphantom{\frac{\text{thermal conduct.}}{\text{electrical conduct.}}} \right\} \text{for many metals}$$

$$\frac{K}{\sigma T} = \text{proportionality constant}$$

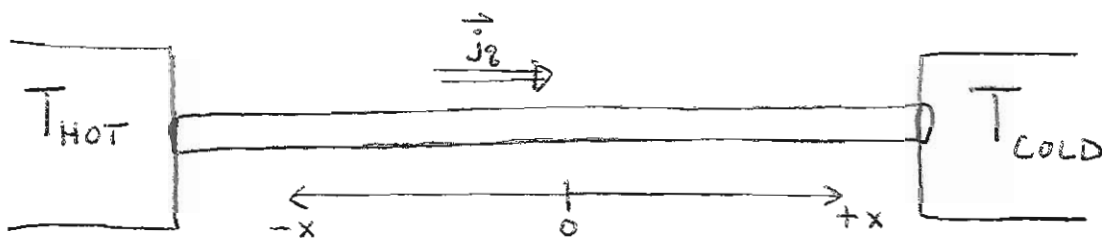
$$\frac{K}{\sigma T} = \text{Lorenz number} = L$$

experimentally,

$$L \approx 2.2 \times 10^{-8} \text{ watt-ohm/K}^2$$

### Drude explanation

assumption: bulk of  $\vec{j}_q$  thermal current in a metal carried by the conduction electrons



$$\frac{dT}{dx} = - \text{(negative)}$$

$$\vec{j}_q = +$$

$$\vec{j}_q = -K \nabla T$$

Goal: is to use Drude model to calculate the Lorenz number

through some basic concepts, one can get:

1-D 
$$j^z = \frac{1}{2} n v [\mathcal{E}(T[x-v\tau]) - \mathcal{E}(T[x+v\tau])]$$

$$= \frac{\text{elec.}}{\text{cm}^3} \times \frac{\text{cm}}{\text{s}} \times \text{energy/electron} \quad \left. \vphantom{\frac{\text{elec.}}{\text{cm}^3}} \right\} \text{dimensional analysis}$$

$$j^z = \frac{\text{energy}}{\text{cm}^2 \cdot \text{s}} = \text{energy flux}$$

$j^z$  can be written as:

$$j^z = -\frac{1}{2} n v \frac{(\Delta T)}{(\Delta x)} [\mathcal{E}(T[x+v\tau]) - \mathcal{E}(T[x-v\tau])]$$

$$= \cancel{n v^2 \tau} \frac{d\mathcal{E}}{dT} \frac{dT}{dx}$$

$$= n v_x^2 \tau \frac{d\mathcal{E}}{dT} \frac{dT}{dx} \quad \text{where } 2v\tau \Rightarrow dx$$

$$j^z = n v_x^2 \tau \frac{d\mathcal{E}}{dT} \left( \frac{-dT}{dx} \right)$$

$(x+v\tau) - (x-v\tau) = 2v\tau$

with 3-D where  $\langle v_x^2 \rangle = \frac{1}{3} \langle v^2 \rangle$

$$j^z = \cancel{\frac{1}{3} n v^2 \tau} \frac{1}{3} n v^2 \tau \frac{d\mathcal{E}}{dT} (-\nabla T)$$

$$\vec{j}^2 = \frac{1}{3} v^2 \tau \left[ \frac{dNE}{dT} / v \right] (-\nabla T)$$

$$= \frac{1}{3} v^2 \tau \left[ \frac{dE/dT}{v} \right] (-\nabla T)$$

$$\vec{j}^2 = \frac{1}{3} v^2 \tau C_v (-\nabla T)$$

compare to

$$\vec{j}^2 = -K(\nabla T)$$

$$\Rightarrow K = \frac{1}{3} v^2 \tau C_v = \frac{1}{3} v \lambda C_v$$

$$\boxed{\lambda = v\tau}$$

$$v^2 = \langle v^2 \rangle$$

= mean square  
electronic speed

Then if we use

$$\tau = \frac{ne^2 \tau}{m}$$

$$\Rightarrow \frac{K}{A} = \frac{\cancel{\frac{1}{3} v^2 \tau C_v}}{\cancel{\frac{ne^2 \tau}{m}}}$$

$$\frac{K}{A} = \frac{1}{3} \frac{mv^2 C_v}{ne^2}$$

classical ideal gas law:  $\Rightarrow C_v = \frac{3}{2} nk_B$

$$\frac{1}{2} mv^2 = \frac{3}{2} k_B T$$

$$\Rightarrow \frac{1}{\sigma} = \frac{1}{3} \frac{2 \left(\frac{3}{2} k_B T\right) \left(\frac{3}{2} n k_B\right)}{n e^2}$$

$$\frac{1}{\sigma} = \frac{3}{2} \left(\frac{k_B}{e}\right)^2 T$$

$$\Rightarrow \text{Lorenze \#} = L = \frac{\kappa}{\sigma T} = \frac{3}{2} \left(\frac{k_B}{e}\right)^2 = 1.11 \times 10^{-8} \text{ watt-ohm/K}^2$$

but Drude had a mistake; he had found

$$\sigma = \frac{n e^2 \tau}{2 m}$$

↑  
erroneous factor of 2

leading to  $\frac{1}{2}$  too small  $\sigma$

leading to 2x larger L ( $\sim 2.22 \times 10^{-8}$  W-ohm/K<sup>2</sup>)

in excellent agreement with experiment

Looking back to

$$\frac{1}{\sigma} = \frac{\frac{1}{3} C_v m \langle v^2 \rangle}{n e^2}$$

actually the true  $C_v$  is about 100x smaller than  $\frac{3}{2} n k_B$   
 and the true  $\langle v^2 \rangle$  is about 100x larger than  $\frac{3 k_B T}{m}$   
 cancelling errors ↑ "lucky" Drude!



# Thermoelectric field

$$\vec{E} = Q \nabla T$$

Q = thermopower

mean electronic velocity

$$v_0 = \frac{1}{2} [v(x-v\tau) - v(x+v\tau)] = -\tau v \frac{dv}{dx}$$

$$= -\tau \frac{d}{dx} \left( \frac{v^2}{2} \right)$$

n 3-D  $\Rightarrow \vec{v}_0 = -\frac{\tau}{6} \frac{d}{dx} (v^2)$

$$= -\frac{\tau}{6} \frac{dv^2}{dT} \nabla T$$

$$\vec{v}_E = -\frac{e\vec{E}}{m} \tau$$

$$\vec{v}_0 + \vec{v}_E = 0$$

$$\vec{E} = -\frac{m}{e} \left( +\frac{1}{6} \right) \frac{dv^2}{dT} \nabla T$$

$$= -\frac{1}{3e} \frac{d}{dT} \left( \frac{mv^2}{2} \right) = -\frac{c_v}{3ne}$$

with  $c_v = \frac{3}{2} nk_B \Rightarrow Q = -\frac{k_B}{2e} = -0.43 \times 10^{-4} \text{ V/K}$

But experimentally observed Q's are  $\sim \underline{\mu\text{V/K}}$  !  
 same factor of 100 error