Review Article

Scanning tunneling microscopy single atom/molecule manipulation and its application to nanoscience and technology

Saw-Wai Hla
Department of Physics and Astronomy, Nanoscale and Quantum Phenomena Institute, Ohio University, Athens, Ohio 45701

(Received 3 June 2005; accepted 6 June 2005; published 18 July 2005)

Single atom/molecule manipulation with a scanning-tunneling-microscope (STM) tip is an innovative experimental technique of nanoscience. Using a STM tip as an engineering or analytical tool, artificial atomic-scale structures can be fabricated, novel quantum phenomena can be probed, and properties of single atoms and molecules can be studied at an atomic level. The STM manipulations can be performed by precisely controlling tip–sample interactions, by using tunneling electrons, or electric field between the tip and sample. In this article, various STM manipulation techniques and some of their applications are described, and the impact of this research area on nanoscience and technology is discussed. © 2005 American Vacuum Society. [DOI: 10.1116/1.1990161]

I. INTRODUCTION

In the mid-20th century, the possibility to image or to see an atom was a matter of great debate. Only after the Nobel-award winning invention of the scanning tunneling microscope (STM) by Binnig and Rohrer in the early 1980s, atomic landscapes of material surfaces could be imaged in real space.1 The operation principle of STM is based on a quantum mechanical phenomenon known as “tunneling.”2 When a sharp needle (tip) is placed less than a 1 nm distance from a conducting material surface (sample) and a voltage is applied between them, the electrons can tunnel between the tip and the sample through the narrow vacuum barrier. Since the tunneling current exponentially varies with the tip–sample distance, a tiny change in the distance less than a fraction of the atomic length can be detected.2 During STM imaging, location of STM tip at the proximity of the surface often causes perturbations due to tip–sample interactions. In a normal imaging mode, these perturbations are not desirable. Since the beginning of the 1990s, these undesired perturbations became one of the most fascinating subjects to be pursued by scientists:3 manipulation of atoms and molecules on surfaces. Using STM manipulation techniques, quantum structures can be constructed on an atom-by-atom basis,4–10 single molecules can be synthesized on a one-molecule-at-a-time basis,11–15 and detailed physical/chemical properties of atoms/molecules, which are elusive to other experimental measurements, can be accessed at an atomic level.16–33 Nowadays, STM is an instrument not only used to see individual atoms by imaging, but also used to touch and take the atoms or to hear their vibration by means of manipulation. In this perspective, STM can be considered as the eyes, hands, and ears of the scientists connecting our macroscopic world to the exciting atomic and nanoscopic world.

II. EXPERIMENTAL SETUP

Manipulation of single atoms and molecules with a STM tip generally requires atomically clean surfaces and an extremility at the tip–sample junction. Therefore, ultrahigh-vacuum environments and low substrate temperatures are favored in most manipulation experiments. Fig. 1 illustrates a custom-built low temperature (LT) STM system capable of performing a variety of manipulation procedures described in this article. This LT STM system, which is built based on the design of Meyer,34 includes a Besco-Beetle type STM scanner35 attached to the base part of a liquid-helium bath cryostat. The system is equipped with a state-of-the-art UHV facilities, sample cleaning and atom/molecule deposition sources. The STM scanner is housed inside two thermal radiation shields, which also act as cryogenic pumps providing a local pressure well below $4 \times 10^{-11}$ Torr. This allows maintaining atomically clean sample condition for a long period of time. Electrochemically etched polycrystalline tungsten wires are used as the tips in most experiments described here. The tip apex is usually prepared by using an in situ tip formation procedure8 described in the next section.

III. PREPARATION FOR MANIPULATION BY TIP CRASH

When a STM tip plunges into a material surface, which is often known as “tip crash,” it causes a crater on the surface. Normally, a tip crash is an undesired event because it can destroy the sample surface. However, the STM tip-crash procedure is useful in several applications including nanoinden- tation to test hardness of material surfaces,36–41 and nanowire...
formation between the tip and sample.\textsuperscript{42–44} Recently, controlled tip-crash procedures are demonstrated to be useful in reshaping the tip apex and extracting individual atoms from the native substrate.\textsuperscript{8}

The shape, structure, and chemical identity of the tip apex play a vital role in STM manipulations. Therefore, formation of an atomically sharp tip apex with known chemical identity is a necessary step for a manipulation experiment. To form the tip apex, the tip is dipped into a metallic substrate for a few nanometers under a high bias condition. The thermal energy produced by the tip crash causes reshaping of the tip apex into a sharper profile. During the process, the tip apex is also coated with the substrate material and hence, its chemical identity can be known.\textsuperscript{8} For extraction of individual atoms from the native substrate, the controlled tip crash is performed under a low tunneling bias condition. After the procedure, scattered individual atoms can be observed near the tip–surface contact area\textsuperscript{8} (Fig. 3). In the case of a tip crash using a silver-coated tungsten tip on a Ag(111) surface, mostly silver atoms are extracted.\textsuperscript{8} These atoms can be used as the basic building blocks of atomistic constructions. If the single atoms can be locally extracted from the native substrate, additional atom deposition is no longer necessary and hence it has an advantage in some nanoscale experiments.

IV. LATERAL MANIPULATION

A STM manipulation procedure to relocate single atoms/molecules across a surface is known as the “lateral manipulation” (LM)\textsuperscript{3,7,9,10,17,34,45–65} (Fig. 4). The first example of LM was demonstrated by Eigler and Schweizer in 1990 by writing the “IBM” company logo with Xe atoms on a Ni(110) surface.\textsuperscript{3} An extremely fine control over the tip–atom–surface interaction is necessary to achieve the atomic-scale precision. A typical LM procedure involves three steps: (1) vertically approaching the tip toward the manipulated atom to increase the tip–atom interaction, (2) scanning the tip parallel to the surface where the atom moves under the influence of the tip, and (3) retracting the tip back to the normal image height thereby the atom is left at the final location on the surface (Fig. 4).

In a LM process, the tunneling resistance (\( R_t \)) is used to indicate the tip–atom distance, and to qualitatively estimate the tip–atom interaction strength.\textsuperscript{9,45} Large and small \( R_t \) correspond to far and close tip–atom distances, or strong and weak tip–atom interactions, respectively. The minimum tip–atom distance required for a manipulation can be determined by measuring a threshold tunneling current \( I_T \) to move an atom at a fixed bias \( I_T \) (Fig. 4(b)).\textsuperscript{9} Figure 4(c) shows a plot of \( I_T \) as a function of applied bias. Here, each data point is determined by plotting a curve, like the one shown in Fig. 4(b). From the slope of this curve, a threshold tunneling resistance required to move the atom can be determined. In the case of a silver atom manipulation on Ag(111), \( R_t = 184\pm8 \Omega \) is necessary.\textsuperscript{9} This \( R_t \) corresponds to a distance of 1.9 Å between the edges of van der Waals radii of tip apex and manipulated atom. Since the atomic orbitals of the tip apex and the manipulated atoms are overlapping at this distance, a weak chemical bond is formed. The attractive force used in the “pulling” manipulation is originated from this chemical nature of interaction.

The nature of atom movements and the type of tip–atom interactions during a LM process can be determined from the STM feedback or tunneling current signals.\textsuperscript{7,9,17,45,47,49,52} The nature of forces involved in a LM process and the mecha-

![Fig. 1. Low-temperature STM for single atom/molecule manipulation. (a)Schematic drawing of STM scanner setup. (b) Image showing a custom-built UHV-LT-STM system at the author’s lab.](image-url)
It vertical force component is initially located directly above the atom and hence, only a
the STM is set in the constant current scanning mode. The tip
moves smoothly across the surface together with the tip. The atom
is virtually bound to or trapped under the tip and it moves smoothly across the surface together with the tip. In the sliding mode, the atom follows the tip due to an attractive tip–atom interaction.

In the pushing mode, a repulsive tip–atom interaction drives the atom to move in front of the tip. This action alerts the STM feedback system to retract the tip in order to maintain the current constant causing an abrupt increase in the tip height. Repeating the sequence provides a saw-tooth-like tip height curve.

In the pushing mode, the tip first climbs up the contour of the atom, and at some point the atom moves away from the tip due to the repulsion. As a result, an abrupt decrease in the tip height occurs and repeating of this sequence provides another saw-tooth-like signal with a reverse form of the pulling curve. The pulling and pushing manipulation signals can be recorded in the constant height scanning mode as well.

The pushing mode described previously are mostly observed when an atom is manipulated along a close-packed row direction (a [110] direction) on face-centered-cubic (fcc) surfaces. Atoms usually prefer to travel along close-packed directions on these surfaces due to a lower diffusion barrier as compared to other surface directions. Sophisticated atom movement mechanisms occur when the atom is manipulated along the directions away from the close packed direction. A silver atom movement along various manipulation directions on a Ag(111) surface is illustrated in Fig. 6. For simplicity, we will describe the manipulation directions by means of deviation angle θ from a surface close-packed direction. Due to the Ag(111) surface symmetry the manipulation paths between θ=0° and 30°, i.e., between the [110] and [211] surface directions (Fig. 6), can reveal the atom movements along any directions on this surface.

At θ=0°, the tip path is along a surface close-packed direction and a typical pulling signal with single lattice-site hops of atom is observed [Fig. 6(b)]. In θ=5° and 10° tip paths, the atom hops hexagonal-close-packed (hcp) and fcc lattice sites upon following the tip producing smaller hopping steps in the manipulation signal [Figs. 6(b) and 6(c)]. The fcc–hcp hopping is induced due to a slight lateral displacement of the tip apex either to the left- or right-hand side of the close-packed row where the atom is traveling. In θ=15° and 20° tip paths, the atom initially follows the tip by preferentially traveling along a surface close-packed direction. However, the atom cannot travel along this path for a longer distance because the tip is moving into other direction away from the atom. At one point, the atom rests for a rela-
tively longer time than the normal hopping time. This is due to a competition between the atom’s preferential movement along the close-packed direction, and its tendency to follow the tip due to the attractive interaction. As the tip scans along the downward slope of the atom, the tip height is reduced resulting in a deeper slope in the manipulation curve [Figs. 6(b) and 6(c)]. This action of the tip increases the lateral force component, $F_L$. When $F_L$ overcomes the hopping barrier, the atom jumps to the adjacent surface close-packed row to follow the tip, which produces an abrupt increase in the tip height. In this tip path, the atom moves by repeating the “move–rest–jump” sequence in an approximate “zigzag” path to follow the tip. In a $\theta=30^\circ$ tip path, the atom smoothly slides the first two neighboring hollow sites located 1.67 Å apart along a [211] direction [Figs. 6(a) and 6(c)]. The third site is located at a relatively far distance (3.33 Å) than the nearest-neighbor distance (2.89 Å) of Ag(111). Instead of traveling directly to the third site, the atom travels approximately in a semicircle path by visiting two adjacent hcp–fcc hollow sites. Thus, all the atom visiting sites in this case are 1.67 Å distance apart to each other. Theoretical simulation reveals that these sites are energetically favored for the atom to move.

The atom manipulations described above correspond to one dimensional tip paths. If LM is performed for a two-dimensional surface area, a series of manipulation signals (either tip height or current signals) can be shown as an image. For manipulation of large molecules, more complex movement behaviors can be observed. Depending on the molecule type and nature of intramolecular bonding, large molecules can have more internal degrees of freedom. This allows large molecules to alter their molecular conformation to adapt the surface potential landscape changes during manipulation. The LM process can be used to probe detailed internal conformation changes of large molecules during their movement across a surface.

Manipulation of single atoms/molecules on surfaces allows construction of various quantum structures and investigation of novel phenomena associated to these structures. By constructing quantum corrals, the standing-electron waves can be systematically studied, electron lifetime inside an artificial structure can be measured, and quantum transport phenomenon can be probed. By manipulating single CO molecules, Heinrich et al. have constructed a molecular computing circuit that can perform logic operations. The atom/molecule manipulation using the STM tip is not only a scientific research but also applicable as an art. To exemplify the variety and novelty of atomistic constructions, a STM manipulation sequence of Ohio University logo (OU), and an “atomic smiley” created by the author are shown in Fig. 7.
V. VERTICAL MANIPULATION

Vertical manipulation (VM) is another useful manipulation technique and it involves the transfer of atoms/molecules between the tip and surface \(^{62,67-73}\). This process is closely related to desorption and subsequent adsorption of atoms/molecules on surfaces. An “atomic switch” realized by repeated transfer of a “Xe” atom between the STM tip and a Ni (110) substrate is the first example of VM. \(^{70}\) The atom/molecule transfer process can be realized by using the electric field between the tip and sample, or by exciting with inelastic tunneling electrons, or by making tip–atom/molecule mechanical contact. The atom/molecule transfer mechanism can be explained by using a double potential well model \(^{72,74}\). At an imaging distance—at which the tip is roughly 6 Å above the surface—the manipulated atom/molecule has two possible stable positions, one at the surface and one at the tip apex. Each position is represented by a potential well and the two potential wells are separated by a barrier \(^{72,74}\). If the tip is positioned above the atom and an electric field is applied, then the shape of the double potential well changes to the one shown with the dashed line in Fig. 8(b). In this case, the barrier between the two wells is reduced and the potential well at the tip apex has a much lower energy. Now, the atom/molecule can easily transfer to the tip. By reversing the bias polarity, the minimum potential well can be shifted to the surface side. Then the dashed curve will reverse 180° from the right to the left, and the atom/molecule can be transferred back to the surface. In the case of VM performed by a tip–atom/molecule mechanical contact, the tip height is reduced until the contact has been achieved. At that distance, the two potential wells overlap and appear as one well \(^{72,74}\). The atom/molecule is then easily transferred to the tip.

In the case of vertical manipulation of CO molecules, a temporary tunneling electron attachment into a 2\(\pi^*\) antibonding state of CO leads to the breaking of the CO–Cu bond first, and the resultant excited CO molecule can either jump to a nearby Cu surface site or to the tip apex. \(^{67,68}\) The CO can be transferred back to the surface using the same process. One useful application of VM is to modify the STM tip. A single atom/molecule tip can be fabricated by deliberately transferring an atom/molecule to the tip apex. \(^{11,12,17,19,32,68}\) This improves the tip sharpness and thus the STM image resolution can be enhanced. \(^{12,17,19,32}\) The chemical identity of the tip apex can also be known from this process. The molecule tips are useful in molecular recognition imaging. For example, CO and oxygen can be distin-
distinguished when a CO tip is used to image. An application is demonstrated in Fig. 9. The VM procedure can also be used to transport the atoms/molecules across a surface. In this case the VM process is analogous to a loading and unloading operation of a crane at a construction site in our macroscopic world.

VI. INELASTIC TUNNELING INDUCED MANIPULATION

Controlled excitations of atoms and molecules can be performed by using inelastic-electron tunneling (IET) induced manipulation processes (Fig. 10). In an IET manipulation, low energy tunneling electrons or holes (if electron tunneling is from the surface to the tip) are injected to the atom/molecule located on a surface by positioning the tip above the target. The tunneling electron energy is transferred to an atom/molecule through a resonance state leading to various excitations. In an IET manipulation process, the maximum tunneling-electron energy can be controlled by adjusting the applied bias. The excitation rate can be varied by changing the tunneling current. If the tunneling current changes, the number of tunneling electrons passing through the atom/molecule will change accordingly and hence, the probability and the rate of excitation will also vary. Two IET processes, single and multiple excitations, can be

Fig. 6. (Color) (a) STM image and a model demonstrate the tip paths and Ag(111) surface geometry. (b) Atom manipulation signals along the tip paths correspond to θ=0°, 5°, 10°, 15°, 20°, 25°, and 30°. (c) Atom movements are demonstrated over an atomically resolved 3D STM image of Ag(111). At θ=5° and 10°, the small steps in manipulation signals are due to fcc–hcp site hopping. At θ=15° and 20°, the periodic downward slopes followed by a tip-height increase is due to the “rest” and “jump” action of the atom. At 30°, the atom slides the first two nearby sites producing two consecutive bumps, and then it travels through a semicycle path to reach the third site. At θ=25°, the manipulation signal reveals both structures of θ=20° and 30°.

Fig. 7. OU logo writing sequence using individual silver atoms on a Ag(111) surface at 6 K (upper) and a three-dimensional representation (middle) (42 nm × 26 nm area, 51 silver atoms are used). “Atomic smiley” image is written by using silver atoms on a Ag(111) surface at 5 K (32 nm diameter).
tron energy transfer is known as the single excitation. An excitation event caused by a single electron energy transfer is known as the single excitation. Several electrons are involved in a multiple excitation process, and it can be explained by using a harmonic oscillator model [Fig. 10(b)]. Here, the electron-energy transfer causes the atom/molecule to excite at an energy level, and subsequent energy transfer by other electrons raises the excited energy level further. The multiple excitation process requires a longer lifetime of the excited state in order to couple the excitations. In the case of molecular excitations; rotational, vibrational, and electronic excitation can be induced. Bond breaking and bond formation between the molecular fragments can also be realized by using IET manipulation. It is possible to determine the number of tunneling electrons involved in an IET process, by using the relation: 

\[ R \propto I^N, \]

where \( R \) is the excitation rate, \( I \) is the tunneling current, and \( N \) is the number of tunneling electrons. In the case of oxygen dissociation on Pt(111), a pioneering work of IET dissociation done by Stipe et al., both single and multiple excitation processes have been successfully demonstrated. Controlled dissociation of polyatomic molecules using tunneling electrons is more complex than that of diatomic species like \( \text{O}_2 \). This is because polyatomic molecules can have a variety of bonds and hence, the tunneling process may involve more than one bond. As an illustration, the dissociation of two C–I bonds of a \( \text{C}_6\text{H}_4\text{I}_2 \) molecule is shown in Fig. 11.

**VII. ELECTRIC FIELD INDUCED MANIPULATION**

The electric field induced manipulation uses the voltage applied between the tip and sample to manipulate single atoms/molecules. By changing the bias polarity, the atom/molecule having a dipole can experience either attractive or repulsive force from the tip. For example, in case of a vertical manipulation of Xe atoms or CO molecules, the changing of the electric field polarity can lead to direct desorption and readsoption of the atom/molecule between the tip and the surface. Smaller molecules, like CO, can be moved laterally across a surface using this manipulation technique. When a high electric field (\( \geq 3 \text{ V} \)) is used, bond breaking of molecules can be performed. Dujardin et al. has first demonstrated dissociation of \( \text{B}_{10}\text{H}_{14} \) and \( \text{O}_2 \) molecules on \( \text{Si}(111) \) using high STM biases (\( \geq 4 \) and \( \geq 6 \text{ V} \), respectively). At the field emission regime where the bias is higher than the work function of the tip, the tip acts as an electron emission gun. In this case, the number of electrons passing through an atom/molecule is no longer controllable.

**VIII. COMBINED USAGE OF STM MANIPULATIONS**

Innovative experimental schemes to investigate specific properties of single atoms/molecules can be developed by combining a variety of STM manipulation procedures. A good example is the induction of single molecule Ullmann reaction on a \( \text{Cu}(111) \) surface where all the basic reactions steps; dissociation, diffusion and association, have been realized by using several manipulation techniques in a step-by-step manner. This chemical reaction sequence results in synthesis of a biphenyl molecule out of two iodobenze. To provide the readers a taste of a combined manipulation experiment, a molecular shooting scheme recently developed by the author is presented. Here, physisorbed sexiphenyl molecules on \( \text{Ag}(111) \) are used as bullets and they are shot toward targets by manipulating with the STM tip. Sexiphenyl is composed of six \( \pi \) rings, which are connected like a linear chain. Sexiphenyl is weakly bound to \( \text{Ag}(111) \) surface, and this makes it difficult to laterally ma-
nipulate the molecule with an atomic precision. During the LM process, the molecule occasionally slides further across the surface after retracting the tip. Using this property of molecule on Ag/H2O\textsubscript{111}/H2O\textsubscript{50}, a molecular shooting scheme has been developed.\textsuperscript{7} To shoot the molecule, it is dragged with the tip toward the long molecular-axis direction for a few nanometers. When the molecule is released by retracting the tip, it continues to propagate up to 30 nm distance on Ag/H2O\textsubscript{111}/H2O\textsubscript{50} surface at 6 K. During this shooting process, the molecular trajectories are not in straight lines because the random electron-standing waves and defects on the surface can cause the molecule to deflect from its original propagation path.

In order to achieve an atomically straight trajectory of the molecules and to probe the mechanism of molecular propagation, a uniform electronic structural environment is necessary. For this, an electron resonator having linear parallel standing-wave front is constructed (Fig. 12). The silver atoms used for the construction are extracted from the native substrate by means of a tip crash procedure described previously.\textsuperscript{8} Shooting sexiphenyl with the STM tip inside this quantum structure provide atomically straight trajectories. To demonstrate the atomic scale control, two sexiphenyl molecules (bullets) and two silver atoms (targets) are positioned at opposite ends of the linear resonator [Fig. 12(a)]. Then one of the molecules is shot toward the target silver atom. As soon as the molecule hits the target, a sexiphenyl–Ag complex is formed [Fig. 12(b)]. This shooting can be performed both at the standing wave minimum and maximum tracks. Thus, uniform electronic structural environment is the key for an atomically controlled shooting process.

It is known that the rings of sexiphenyl are alternately twisted in the gas phase with a torsional angle between 20° and 40° due to the steric repulsion. Sexiphenyl on Ag(111) has an alternately twisted ring configuration.\textsuperscript{17} To determine the ring twisting angle and internal conformation changes during its movement, a controlled LM process is performed inside the electron resonator. The tip is positioned at an edge of a sexiphenyl ring during LM to detect the ring movement. The manipulation signal reveals periodic low–high tip height peaks with hcp–fcc lattice distance of Ag(111) [Fig. 12(c)]. Because of the constant current scanning mode, the tip needs to retract in order to maintain the current constant when the ring edge is tilted upward. Thus, the higher peak corresponds to the upward tilted ring edge configuration. When the p-ring edge is tilted downward, a low-peak signal is produced. This reveals that the sexiphenyl moves across the surface by flipping its alternate ring twist to and fro, like a caterpillar moving on the ground. The molecule shooting mechanism is triggered by the mechanical...
energy stored in its twisted π-ring configuration during dragging with the STM tip.7 From the manipulation geometry, the torsional angle of sexiphenyl on Ag(111) can be determined as 11.4°.17 This combined manipulation experiment set an example where the experiments can be conducted within a few tens of nm square area and the single molecule level information, which can not be accessed by other experimental techniques, can be obtained.

IX. OUTLOOK

The progress in manipulation techniques has not only extended the application of STM but also created a new field of nanoscale engineering. Construction of atomic scale structures, such as quantum corrals and electron resonators, allows the study of quantum phenomena locally.4,5,7,9,10,66 Inducing chemical reactions at single molecule level with the STM tip gives a distinctive insight to the reaction mechanism at a fundamental level.11,13,14 Chemical reactions like the Ullmann reaction can be verified on an entirely new platform.12,13 Novel chemical reaction pathways can be discovered.11 Such manipulation experiments are fascinating and, without a doubt, greatly contribute to the advancement of science. Most STM atom/molecule manipulations require sophisticated instruments to achieve a precise control at the tip–sample junction. Thus only a few research groups in the world have been able to use such manipulation techniques as compared to a large number of research groups using STM imaging and spectroscopy methods. However, this trend is gradually changing in recent years. Room temperature manipulations have been able to perform on large molecules.86–90 The atom manipulation process can also be automated, which increases the speed of atomistic construction substantially. With all the exciting advances in manipulation techniques, the STM atom/molecule manipulation will continue to be a powerful analytical technique of nanoscience, just as the STM imaging and spectroscopy. STM imaging has greatly contributed to surface science and semiconductor industry. It is certain that the STM manipulation will follow a similar trend coupled with equally outstanding contributions to nanoscience research and future nanotechnology.

ACKNOWLEDGMENTS

The support provided by the U.S. Department of Energy Grant No. DE-FG02-02ER46012, the NSF Grant No. DMR 0304314, and the Ohio University CMSS Program are gratefully acknowledged.
